



THE OHIO STATE UNIVERSITY

“Total Mass Accounting in Advanced Liquid-Fueled Reactors”

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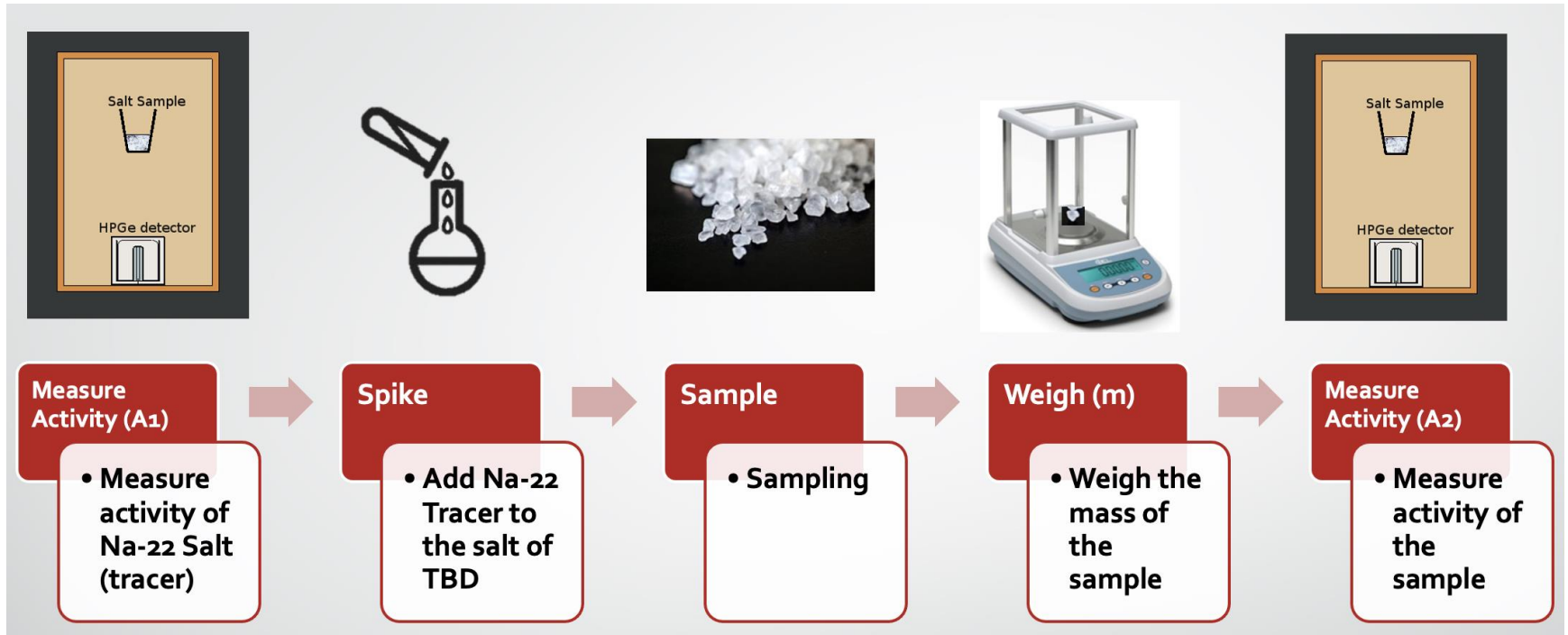


Objectives

The objective of this research is to validate a radioactive tracer dilution (RTD) method for the irradiated fuel-bearing molten salt mass determination to evaluate the possibilities of its deployment in NMA scenarios, e.g., in molten salt loop in LFMSRs.



The principle of RTD



$$Mass_{tot} = \left(\frac{A_1}{A_2} \right) * m_{sample}$$



Radioactive tracer selection:

^{22}Na was selected as the first radioactive tracer to test the RTD concept because –

- It undergoes beta+ decay (non-fission product characteristic)
- known chemical compatibility with actinides and fission products in molten chloride and fluoride salts
- Availability and the half life (2.6y) for handling
- Emits 1274.54 keV gamma-ray (99.94%), high enough to be outside of the Compton plateau of many fission products' gamma-rays in a gamma energy spectrum
- Identified overlapped peak at 1274.43 keV from ^{154}Eu



Experimental Method (2017-2021)

Step 1

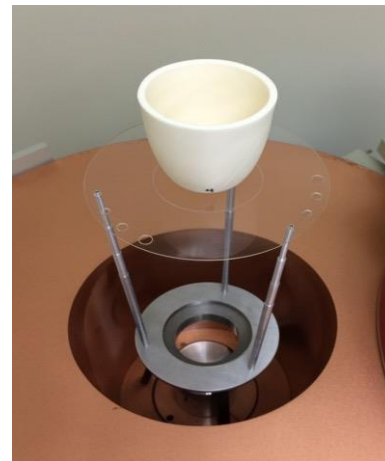
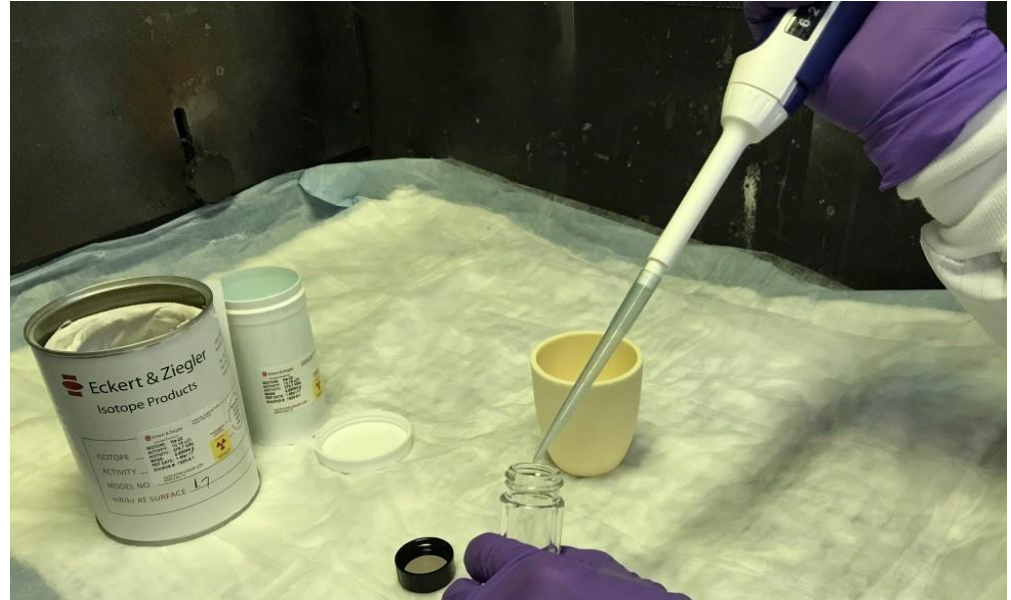
- Add Tracers to Empty Crucible
 - 1.29 μCi Na-22
 - 3.1 μCi Eu-154
 - 10.06 μCi Cs-137

Step 2

- Add LiCl-KCl Salt to crucible and melt to form one solid piece

Step 3

- Measure Activity of Tracer+ Salt solid





Prior work: Proof-of-concept with the clean salt (contd.)

Step 4

- Re-melt, solidify, and partition salt into separate crucibles

Step 5

- Re-melt, re-solidify each of crucibles

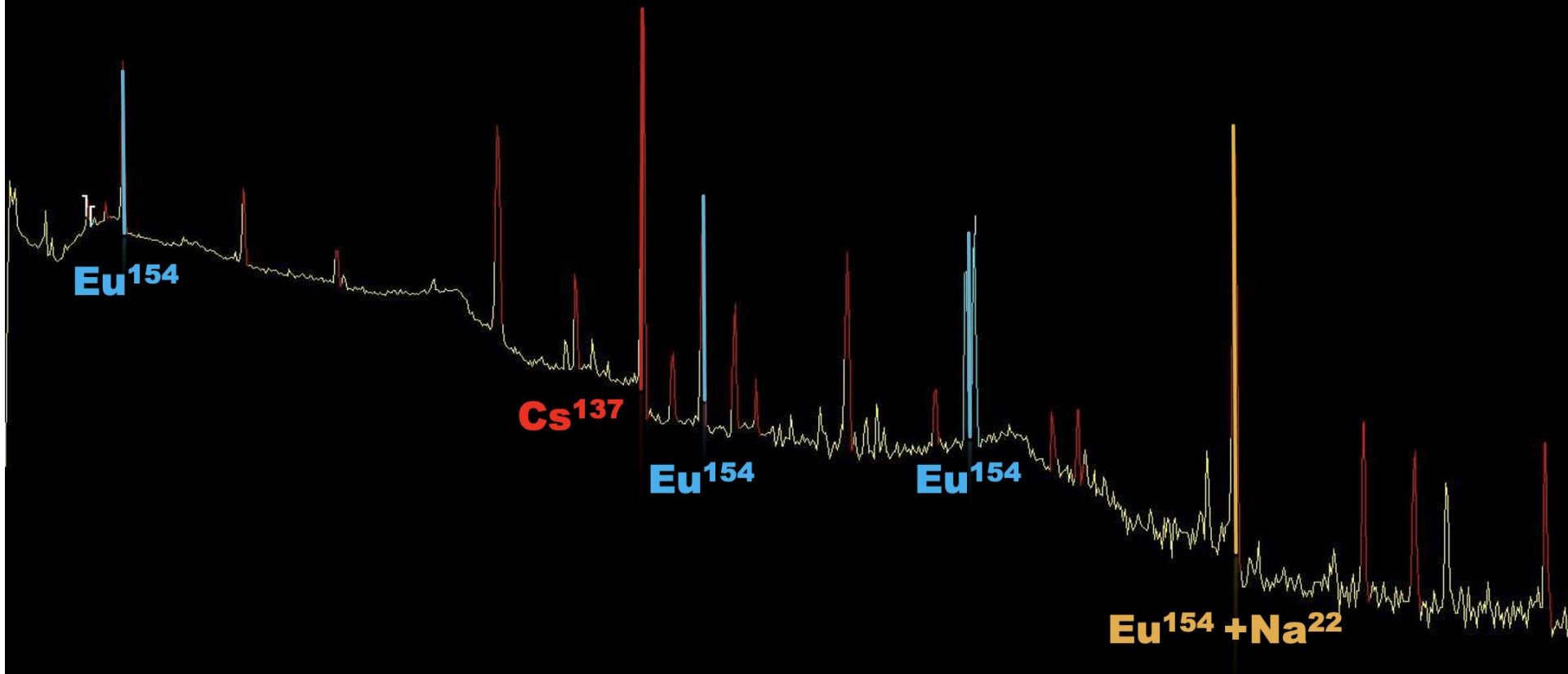
Step 6

- Measure activity of Tracer+ Salt in separate crucibles





Gamma Spectrum from 5 g of salt with ^{137}Cs , ^{154}Eu and ^{22}Na



Acquired at OSU Nuclear Reactor Lab



Challenges: Off-gas Constituents

Table 1

Potential species in the cover gas of an MSR.

Type of cover gas constituent	Example species
Mists, aerosols, and particles	Salt residues, graphite debris for graphite-moderated fluoride systems, corrosion products, and noble metals (e.g., Ru, Pd, Rh)
Gases and volatile species	^3HF , HF , F_2 , Cl_2 , Br_2 , I_2 , Ar, interhalogens (e.g., ICl , IF_5 , IF_7), volatile halides, and the decay products (e.g., Cs, Ba, Rb, Sr, La, Br, I, Se, Te) (Ostvald et al., 2009)
Tritium	$^3\text{H}_{2(\text{g})}$, $^3\text{HH}_{(\text{g})}$, $^3\text{HF}_{(\text{g})}$, $^3\text{HF}_{(\text{l})}$, and possibly $^3\text{HHO}_{(\text{g})}$ and/or $^3\text{H}_2\text{O}_{(\text{g})}$
Short-lived fission gases and their daughters	$^{139}\text{Xe } t_{1/2} = 39.5 \text{ s}$, $^{137}\text{Xe } t_{1/2} = 3.83 \text{ min}$, $^{135\text{m}}\text{Xe } t_{1/2} = 15.3 \text{ min}$, $^{135}\text{Xe } t_{1/2} = 9.1 \text{ h}$, $^{133\text{m}}\text{Xe } t_{1/2} = 2.19 \text{ d}$, $^{133}\text{Xe } t_{1/2} = 5.25 \text{ d}$, $^{90}\text{Kr } t_{1/2} = 32.3 \text{ s}$, $^{89}\text{Kr } t_{1/2} = 3.18 \text{ min}$, $^{88}\text{Kr } t_{1/2} = 2.84 \text{ h}$
Longer-lived radionuclides	$^{129}\text{I } t_{1/2} = 1.57 \times 10^7 \text{ y}$, $^{79}\text{Se } t_{1/2} = 6.5 \times 10^4 \text{ y}$, $^{85}\text{Kr } t_{1/2} = 10.7 \text{ y}$, $^{36}\text{Cl } t_{1/2} = 3 \times 10^5 \text{ y}$

Source: Andrews, Hunter et al. "Review of molten salt reactor off-gas management considerations." *Nuclear Engineering and Design* **385** (2021): 111529.



- Estimated fission products activities from spreadsheet calculations
- Serpent and MCNP–ORIGEN are being requested from RSICC

INPUT PARAMETERS		Description
$\phi_{th} \text{ (cm}^{-2} \cdot \text{s}^{-1}\text{)}$	1E+13	Thermal Neutron Flux in CIF
$m_{U235} \text{ (g)}$	0.01	mass of U-235
$t_{irr} \text{ (h)}$	1	irradiation time
$t_d \text{ (h)}$	24	decay time after irradiation
$A_{Na22,0} \text{ (Ci)}$	1.00E-03	initial ^{22}Na Activity



Fission Product Activity

Decay Time: 10 h

		Fission Product Activity (μCi)	
	Fission product, half-life, main gamma energy	at $t_{irr} + t$ ($t = 0$)	at $t_{irr} + t$ ($t = t_d$); see B9 for t_d
	¹³⁷ Cs; 30.07 years; 662 keV	6.65	6.65
	¹⁵⁴ Eu; 8.59 years; 1274 keV, 723 keV, 123 keV	0.00	0.00
	¹⁵⁵ Eu; 4.76 years; 86.5 keV, 105.3 keV	0.21	0.21
Krypton	⁸⁵ Kr; 10.7 y; 514 keV	0.86	0.86
	^{85m} Kr; 4.48 h; 151 keV	7.59E+04	1.62E+04
	⁸⁸ Kr; 2.84 h; 2392 keV	3.14E+05	2.73E+04
	⁸⁹ Kr; 3.15 min; 221 keV	1.87E+06	0.00
	⁹⁰ Kr; 32.32 s; 1118.7 keV	2.05E+06	0.00E+00
Iodine	¹²⁹ I; 1.57×10^7 y; 39.578 keV	0.00	0.00
	¹³¹ I; 8 days; 364 keV	4.20E+03	4.05E+03
	¹³³ I; 20.8 h; 530 keV	8.77E+04	6.29E+04
	¹³⁵ I; 6.57 h; 1260 keV, 1132 keV	2.60E+05	9.05E+04
Xenon	^{131m} Xe; 11.934 d; 163.93 keV	30.74	3.00E+01
	¹³³ Xe; 5.25 d; 81 keV	1.47E+04	1.39E+04
	^{133m} Xe; 2.19 d;	1.01E+03	8.82E+02
	¹³⁵ Xe; 9.14 h; 250 keV	1.96E+05	9.19E+04
	^{135m} Xe; 15.3 min; 250 keV	4.63E+05	7.25E-07
	¹³⁷ Xe; 3.818 min; keV	2.48E+06	0.00
	¹³⁹ Xe; 39.68 s; 218.59 keV	2.09E+06	0.00
	⁹⁰ Sr; 28.8 years	6.40	6.40E+00
	⁹⁵ Zr; 64 days; 724 keV, 757 keV	1.19E+03	1.19E+03
	⁹⁵ Nb; 35 days; 766 keV	2.18E+03	2.16E+03
	TOTAL ACTIVITY (μCi)	9.91E+06	3.11E+05



Fission Product Activity

Decay Time: 100 h

		Fission Product Activity (μCi)	
	Fission product, half-life, main gamma energy	at $t_{\text{irr}} + t$ ($t = 0$)	at $t_{\text{irr}} + t$ ($t = t_d$); see B9 for t_d
	^{137}Cs ; 30.07 years; 662 keV	6.65	6.65
	^{154}Eu ; 8.59 years; 1274 keV, 723 keV, 123 keV	0.00	0.00
	^{155}Eu ; 4.76 years; 86.5 keV, 105.3 keV	0.21	0.21
Krypton	^{85}Kr ; 10.7 y; 514 keV	0.86	0.86
	$^{85\text{m}}\text{Kr}$; 4.48 h; 151 keV	7.59E+04	1.45E-02
	^{88}Kr ; 2.84 h; 2392 keV	3.14E+05	7.89E-06
	^{89}Kr ; 3.15 min; 221 keV	1.87E+06	0.00
	^{90}Kr ; 32.32 s; 1118.7 keV	2.05E+06	0.00E+00
Iodine	^{129}I ; 1.57×10^7 y; 39.578 keV	0.00	0.00
	^{131}I ; 8 days; 364 keV	4.20E+03	2.93E+03
	^{133}I ; 20.8 h; 530 keV	8.77E+04	3.13E+03
	^{135}I ; 6.57 h; 1260 keV, 1132 keV	2.60E+05	6.81E+00
Xenon	$^{131\text{m}}\text{Xe}$; 11.934 d; 163.93 keV	30.74	2.41E+01
	^{133}Xe ; 5.25 d; 81 keV	1.47E+04	8.49E+03
	$^{133\text{m}}\text{Xe}$; 2.19 d;	1.01E+03	2.69E+02
	^{135}Xe ; 9.14 h; 250 keV	1.96E+05	9.98E+01
	$^{135\text{m}}\text{Xe}$; 15.3 min; 250 keV	4.63E+05	4.12E-113
	^{137}Xe ; 3.818 min; keV	2.48E+06	0.00
	^{139}Xe ; 39.68 s; 218.59 keV	2.09E+06	0.00
	^{90}Sr ; 28.8 years	6.40	6.40E+00
	^{95}Zr ; 64 days; 724 keV, 757 keV	1.19E+03	1.14E+03
	^{95}Nb ; 35 days; 766 keV	2.18E+03	2.01E+03
	TOTAL ACTIVITY (μCi)	9.91E+06	1.81E+04



Fission Product Activity

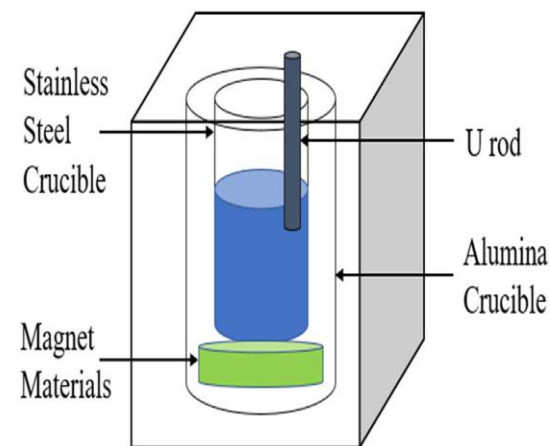
Decay Time: 3 months

		Fission Product Activity (μCi)	
	Fission product, half-life, main gamma energy	at $t_{irr} + t$ ($t = 0$)	at $t_{irr} + t$ ($t = t_d$); see B9 for t_d
	¹³⁷ Cs; 30.07 years; 662 keV	6.65	6.61
	¹⁵⁴ Eu; 8.59 years; 1274 keV, 723 keV, 123 keV	0.00	0.00
	¹⁵⁵ Eu; 4.76 years; 86.5 keV, 105.3 keV	0.21	0.20
Krypton	⁸⁵ Kr; 10.7 y; 514 keV	0.86	0.84
	^{85m} Kr; 4.48 h; 151 keV	7.59E+04	0.00
	⁸⁸ Kr; 2.84 h; 2392 keV	3.14E+05	0.00
	⁸⁹ Kr; 3.15 min; 221 keV	1.87E+06	0.00
	⁹⁰ Kr; 32.32 s; 1118.7 keV	2.05E+06	0.00
Iodine	¹²⁹ I; 1.57×10^7 y; 39.578 keV	0.00	0.00
	¹³¹ I; 8 days; 364 keV	4.20E+03	1.58E+00
	¹³³ I; 20.8 h; 530 keV	8.77E+04	0.00
	¹³⁵ I; 6.57 h; 1260 keV, 1132 keV	2.60E+05	0.00
Xenon	^{131m} Xe; 11.934 d; 163.93 keV	30.74	1.53E-01
	¹³³ Xe; 5.25 d; 81 keV	1.47E+04	8.62E-02
	^{133m} Xe; 2.19 d;	1.01E+03	0.00
	¹³⁵ Xe; 9.14 h; 250 keV	1.96E+05	0.00
	^{135m} Xe; 15.3 min; 250 keV	4.63E+05	0.00
	¹³⁷ Xe; 3.818 min; keV	2.48E+06	0.00
	¹³⁹ Xe; 39.68 s; 218.59 keV	2.09E+06	0.00
	⁹⁰ Sr; 28.8 years	6.40	6.36E+00
	⁹⁵ Zr; 64 days; 724 keV, 757 keV	1.19E+03	4.44E+02
	⁹⁵ Nb; 35 days; 766 keV	2.18E+03	3.57E+02
	TOTAL ACTIVITY (μCi)	9.91E+06	8.17E+02



Preparation of NaCl-KCl- UCl_3 salt for irradiation (UoU)

- 13.8 g of MgCl_2 -KCl- UCl_3 (DU) fuel salt has been prepared at Nuclear PyroMetallurgy Laboratory, the University of Utah
- High purity (99.99%) MgCl_2 and KCl was acquired through commercial vendors and mixed with a 1:1 molar ratio.
- UCl_3 was synthesized by using DU metal rod and FeCl_2 in MgCl_2 -KCl at $\sim 700^\circ\text{C}$. Salt samples were taken and measured by ICP-MS, the UCl_3 concentration was determined to be 15.15 wt%. The FeCl_2 concentration is 0.045 wt%.
- U-235 concentration is **TBD**
- MgCl_2 -KCl- UCl_3 salt will be packed in an argon glovebox and shipped to OSU by a commercial carrier

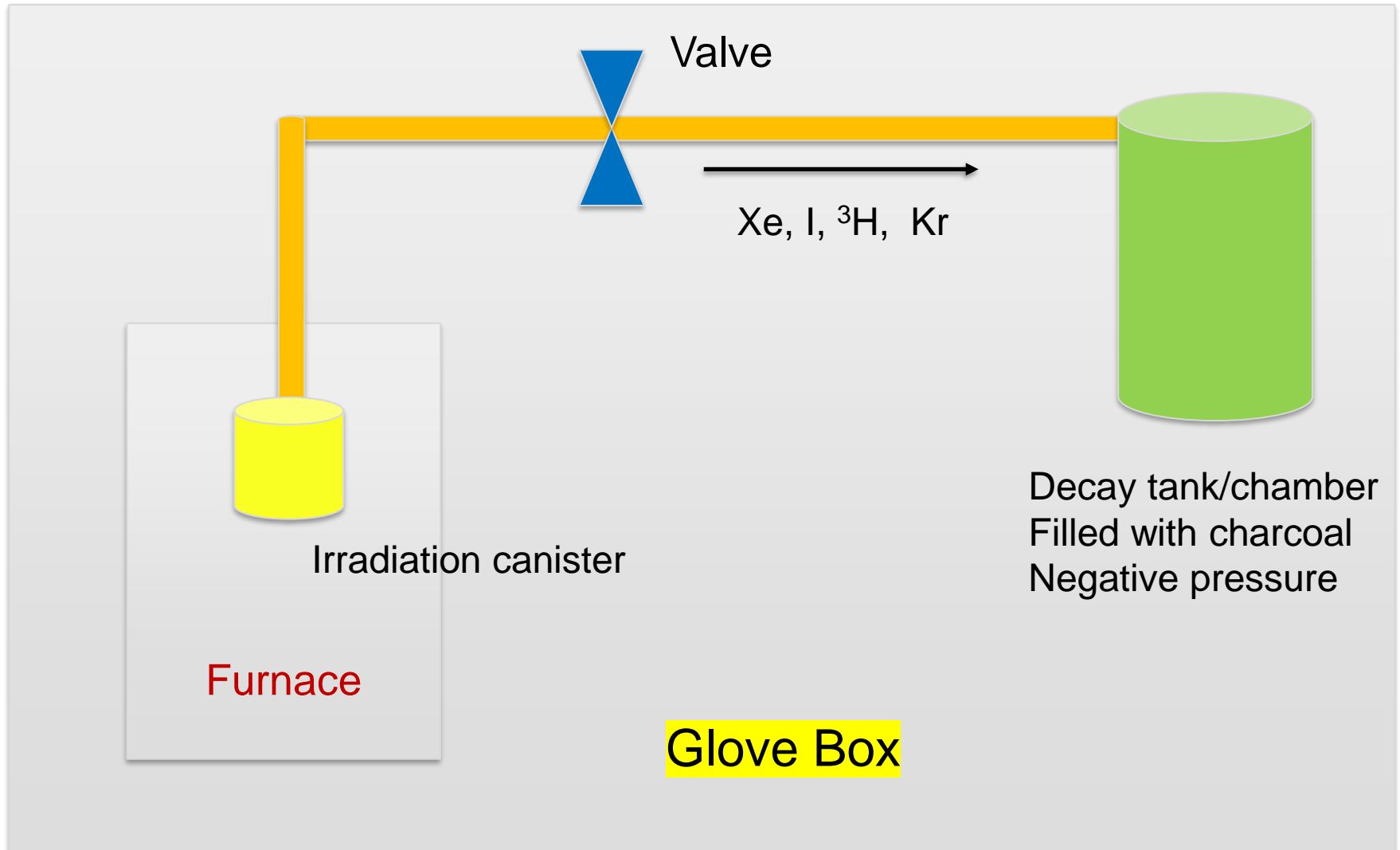


Schematic for preparing UCl_3 salt from DU metal rod and NaCl-KCl- FeCl_2

Huan Zhang et al 2021, J. Electrochem. Soc. 168 056521

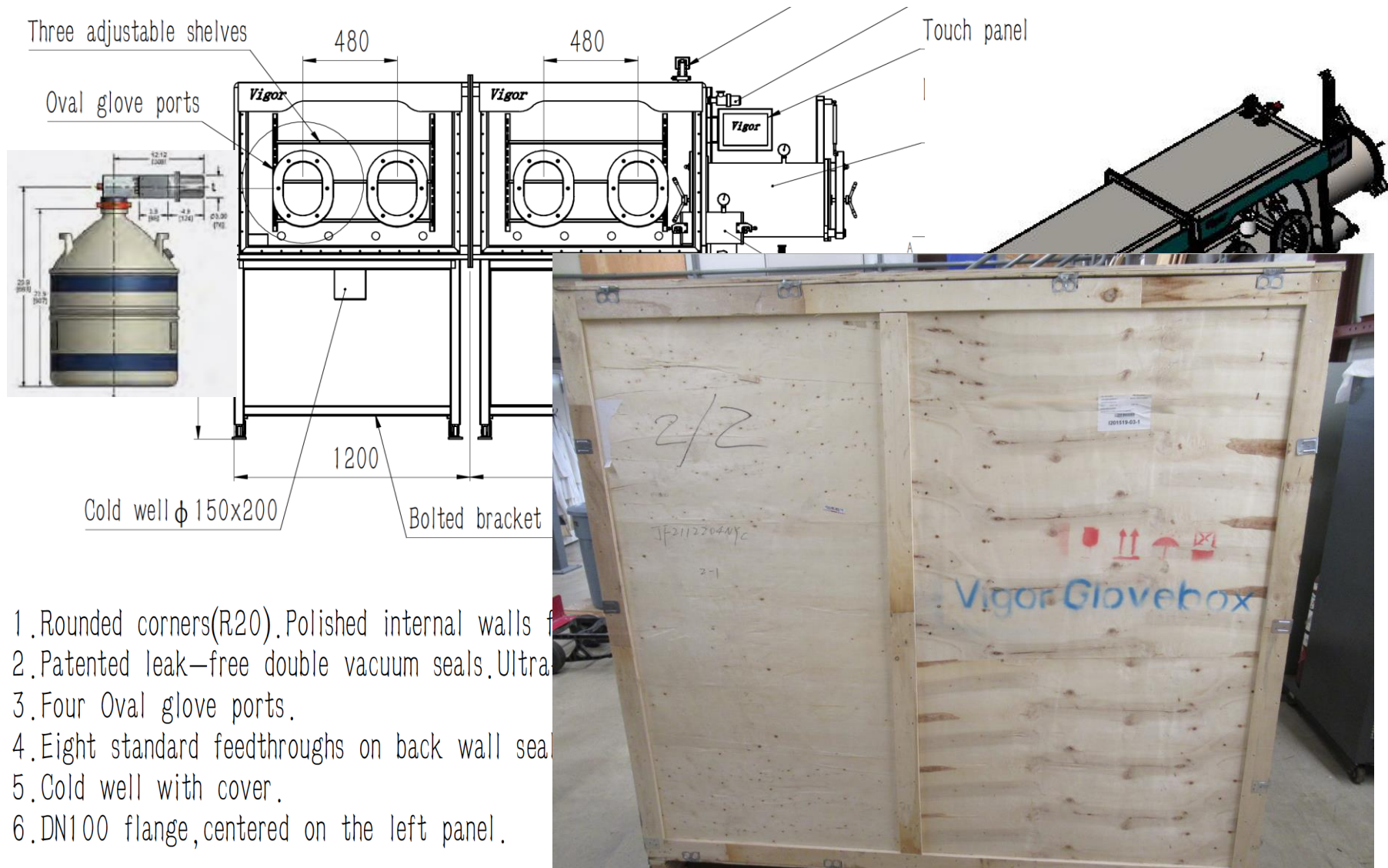


Irradiated salt process and measurement plan





Order placed for a customized glovebox

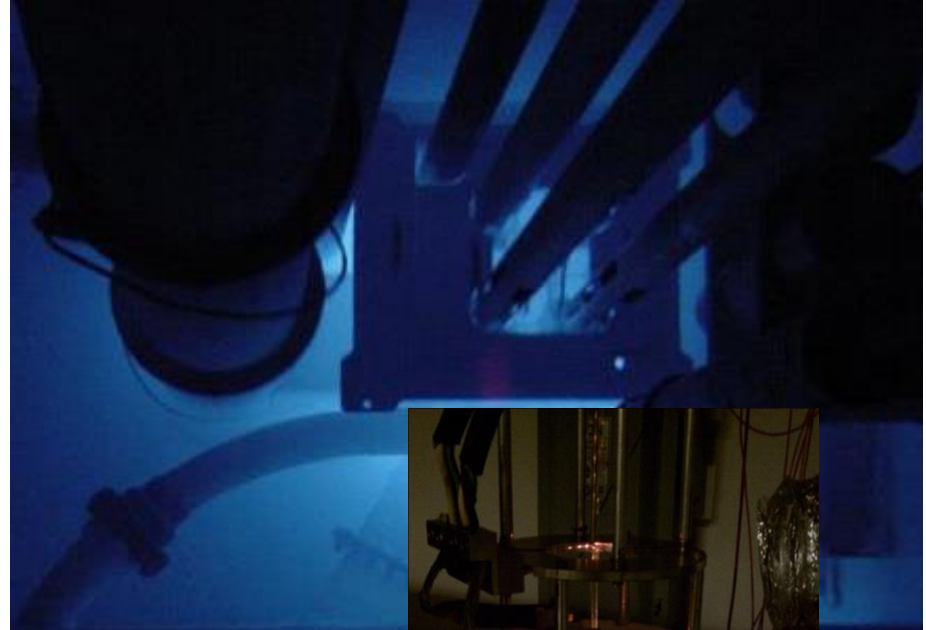


1. Rounded corners(R20).Polished internal walls
2. Patented leak-free double vacuum seals.Ultra
3. Four Oval glove ports.
4. Eight standard feedthroughs on back wall seal
5. Cold well with cover.
6. DN100 flange,centered on the left panel.



Phase 1 Experiment:

- Prepare $\text{UCl}_3 + \text{MgCl}_2 + \text{KCl}$ with DU, ~ mg of U-235,
- Add $^{22}\text{NaCl}$ tracer
- Melted and sealed in irradiation cell
- Irradiate at OSU Research Reactor for 2 hours
- Decay
- Survey the activity and measure the gamma-spectrum (without opening the container)





Estimated Dose Rates for 10 grams of Fuel Salt Mixture:

4 g MgCl₂, 4 g KCl, and 2 g UCl₃

Irradiation Facility: CIF

Flux: 1.6×10^{13} n/cm²/s

Thermal to Fast ratio: 1.4545

Irradiation time: 1 hour

Isotope	Mass (grams)	Distance from source	Time to Decay (after 1 hour irradiation)	Gamma Dose rate (mrem/hr)	Beta Dose rate (mrem/hr)	Total Dose rate (mrem/hr)
Na-24	1.021	1 meter	64 hours	4.18E-02	6.01E-01	6.43E-01
K-42	2.098			2.78E-01	5.41E+01	5.44E+01
S-35 (from Cl)	5.498			0.00E+00	1.52E+02	1.52E+02
P-32 (from Cl)				0.00E+00	9.27E-01	9.27E-01
Na-24	1.021	10 cm	64 hours	4.18E+00	6.01E+01	6.43E+01
K-42	2.098			2.78E+01	5.41E+03	5.44E+03
S-35 (from Cl)	5.498			0.00E+00	1.52E+04	1.52E+04
P-32 (from Cl)				0.00E+00	9.27E+01	9.27E+01
Na-24	1.021	1 meter	7 days	3.41E-04	4.91E-03	5.25E-03
K-42	2.098			8.00E-04	1.56E-01	1.57E-01
S-35 (from Cl)	5.498			0.00E+00	1.47E+02	1.47E+02
P-32 (from Cl)				0.00E+00	7.51E-01	7.51E-01
Na-24	1.021	10 cm	7 days	3.41E-02	4.91E-01	5.25E-01
K-42	2.098			8.00E-02	1.56E+01	1.57E+01
S-35 (from Cl)	5.498			0.00E+00	1.47E+04	1.47E+04
P-32 (from Cl)				0.00E+00	7.51E+01	7.51E+01

Isotope	Mass (grams)	Distance from source	Time to Decay (after 1 hour irradiation)	Gamma Dose rate (mrem/hr)	Beta Dose rate (mrem/hr)	Total Dose rate (mrem/hr)
Cs-137	10 mg U-235	30 cm	64 hours	2.84E-03	2.08E-01	2.10E-01
Eu-154				0.00E+00	0.00E+00	0.00E+00
Eu-155				1.48E-05	6.19E-03	6.20E-03
Kr-85				1.57E-06	2.79E-02	2.79E-02
Kr-85m				6.78E-04	1.18E-01	1.19E-01
Kr-88				5.43E-05	1.48E-03	1.53E-03
Kr-89				0.00E+00	0.00E+00	0.00E+00
Kr-90				0.00E+00	0.00E+00	0.00E+00
I-133				1.06E+01	7.22E+02	7.33E+02
I-131				2.19E+01	2.16E+03	2.18E+03
I-135				1.04E-03	3.36E-02	3.46E-02
Xe-133				1.18E+00	3.20E+02	3.21E+02
Xe-135				3.22E-01	4.74E+01	4.78E+01
Xe-135m				4.30E-74	3.74E-72	3.79E-72
Xe-137				6.99E-301	1.57E-298	1.58E-298
Xe-139				N/A	0.00E+00	N/A
Sm-149				N/A	0.00E+00	N/A
Sr-90				N/A	3.69E+01	N/A
Zr-95				5.98E-01	3.58E+01	3.64E+01
Nb-95				1.10E+00	6.41E+01	6.52E+01
Se-79				#N/A	0.00E+00	#N/A
Total Dose				3.57E+01	3.39E+03	3.38E+03

Cs-137	10 mg U-235	10 cm	64 hours	2.56E-02	1.87E+00	1.89E+00
Eu-154				0.00E+00	0.00E+00	0.00E+00
Eu-155				1.33E-04	5.57E-02	5.58E-02
Kr-85				1.41E-05	2.51E-01	2.51E-01
Kr-85m				6.10E-03	1.06E+00	1.07E+00
Kr-88				4.89E-04	1.33E-02	1.38E-02
Kr-89				0.00E+00	0.00E+00	0.00E+00
Kr-90				0.00E+00	0.00E+00	0.00E+00
I-133				9.53E+01	6.50E+03	6.60E+03
I-131				1.97E+02	1.94E+04	1.96E+04
I-135				9.33E-03	3.02E-01	3.11E-01
Xe-133				1.06E+01	2.88E+03	2.89E+03
Xe-135				2.90E+00	4.27E+02	4.30E+02
Xe-135m				3.87E-73	3.37E-71	3.41E-71
Xe-137				6.29E-300	1.42E-297	1.42E-297
Xe-139				N/A	0.00E+00	N/A
Sm-149	N/A	0.00E+00	N/A			
Sr-90	N/A	3.32E+02	N/A			
Zr-95	5.38E+00	3.22E+02	3.28E+02			
Nb-95	9.93E+00	5.77E+02	5.87E+02			
Se-79	N/A	0.00E+00	N/A			
Total Dose				3.21E+02	3.05E+04	3.05E+04



Estimated Dose Rates for 3 grams of Fuel Salt Mixture:

1 g MgCl₂, 1 g KCl, and 1 g UCl₃

Irradiation Facility: CIF

Flux: 1.6×10^{13} n/cm²/s

Thermal to Fast ratio: 1.4545

Irradiation time: 1 hour

Isotope	Mass (grams)	Distance from source	Time to Decay (after 1 hour irradiation)	Gamma Dose rate (mrem/hr)	Beta Dose rate (mrem/hr)	Total Dose rate (mrem/hr)
Na-24	0.3934	1 meter	64 hours	1.05E-02	1.51E-01	1.61E-01
K-42	0.5244			6.95E-02	1.35E+01	1.36E+01
S-35 (from Cl)	1.0822			0.00E+00	4.23E+01	4.23E+01
P-32 (from Cl)	1.0822			0.00E+00	2.62E+00	2.62E+00
Na-24	0.3934	10 cm	64 hours	1.05E+00	1.51E+01	1.61E+01
K-42	0.5244			6.95E+00	1.35E+03	1.36E+03
S-35 (from Cl)	1.0822			0.00E+00	4.23E+03	4.23E+03
P-32 (from Cl)	1.0822			0.00E+00	2.62E+02	2.62E+02
Na-24	0.3934	1 meter	7 days	8.56E-05	1.23E-03	1.32E-03
K-42	0.5244			2.00E-04	3.89E-02	3.91E-02
S-35 (from Cl)	1.0822			0.00E+00	4.09E+01	4.09E+01
P-32 (from Cl)	1.0822			0.00E+00	2.12E+00	2.12E+00
Na-24	0.3934	10 cm	7 days	8.56E-03	1.23E-01	1.32E-01
K-42	0.5244			2.00E-02	3.89E+00	3.91E+00
S-35 (from Cl)	1.0822			0.00E+00	4.09E+03	4.09E+03
P-32 (from Cl)	1.0822			0.00E+00	2.12E+02	2.12E+02

Isotope	Mass (grams)	Distance from source	Time to Decay (after 1 hour irradiation)	Gamma Dose rate (mrem/hr)	Beta Dose rate (mrem/hr)	Total Dose rate (mrem/hr)
Cs-137	1.4 mg U-235	30 cm	64 hours	3.82E-04	2.79E-02	2.83E-02
Eu-154				0.00E+00	0.00E+00	0.00E+00
Eu-155				0.00E+00	0.00E+00	0.00E+00
Kr-85				1.74E-07	3.10E-03	3.10E-03
Kr-85m				9.47E-05	1.65E-02	1.66E-02
Kr-88				7.59E-06	2.07E-04	2.14E-04
Kr-89				0.00E+00	0.00E+00	0.00E+00
Kr-90				0.00E+00	0.00E+00	0.00E+00
I-133				6.65E-01	4.53E+01	4.60E+01
I-131				1.47E-01	1.45E+01	1.46E+01
I-135				1.45E-04	4.70E-03	4.84E-03
Xe-133				1.66E-01	4.49E+01	4.50E+01
Xe-135				4.52E-02	6.65E+00	6.70E+00
Xe-135m				6.01E-75	5.24E-73	5.30E-73
Xe-137				9.80E-302	2.21E-299	2.22E-299
Xe-139				N/A	0.00E+00	N/A
Sm-149				N/A	0.00E+00	N/A
Sr-90				N/A	2.79E-02	N/A
Zr-95				5.98E-01	3.58E+01	3.64E+01
Nb-95				1.54E-01	8.96E+00	9.12E+00
Se-79				N/A	0.00E+00	N/A
Total Dose				1.78E+00	1.56E+02	1.58E+02

Cs-137	1.4 mg U-235	10 cm	64 hours	0.00E+00	0.00E+00	0.00E+00
Eu-154				0.00E+00	0.00E+00	0.00E+00
Eu-155				5.98E+00	4.08E+02	4.14E+02
Kr-85				1.32E+00	1.30E+02	1.32E+02
Kr-85m				1.31E-03	4.23E-02	4.36E-02
Kr-88				1.49E+00	4.04E+02	4.05E+02
Kr-89				4.07E-01	5.99E+01	6.03E+01
Kr-90				5.41E-74	4.71E-72	4.77E-72
I-133				8.82E-301	1.99E-298	2.00E-298
I-131				N/A	0.00E+00	N/A
I-135				N/A	0.00E+00	N/A
Xe-133				N/A	2.51E-01	N/A
Xe-135				5.38E+00	3.22E+02	3.28E+02
Xe-135m				1.39E+00	8.07E+01	8.21E+01
Xe-137				N/A	0.00E+00	N/A
Xe-139				0.00E+00	0.00E+00	0.00E+00
Sm-149				0.00E+00	0.00E+00	0.00E+00
Sr-90	5.98E+00	4.08E+02	4.14E+02			
Zr-95	1.32E+00	1.30E+02	1.32E+02			
Nb-95	1.31E-03	4.23E-02	4.36E-02			
Se-79	1.49E+00	4.04E+02	4.05E+02			
Total Dose				1.60E+01	1.41E+03	1.42E+03



Sample Vial – 20 mm (10 mL capacity)

- Quartz (Fused Silica)
- Aluminum Foil
- Aluminum Crimp Cap
- Grafoil gasket

External Containment Vessel (70 mL capacity)

- Aluminum Body (Al-1100)
- Grafoil gasket



** Quartz containment vessel
available smaller samples



Aluminum Crimp Seals

*Max Pressure: 40-55 psi
Max Temp: 550 C

*Pressure increase of 1 mL (4e-5 moles)
of argon due to temperature rise from
20°C to 150°C during irradiation
estimated to be 21 psi (~1.4 atm) or less.

Based on ideal Gas law: $PV=nRT$



11 mm Solid Cap

Gaskets

- **MicaTherm**
 - Max Temp: 900 C (in Oxygen)
- **Flexible Graphite**
 - Max Temperature: 450 C (in Oxygen)
2980 C (in Argon)
 - Max Pressure: 1450-2900 psi





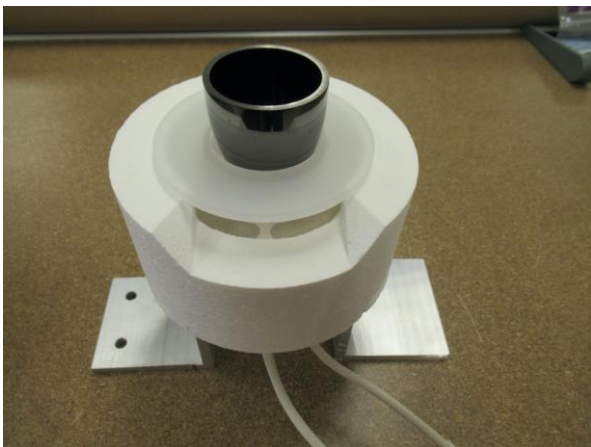
Containment System for Heating and Off-gassing



Ultralite Kiln – (Modified in-house)
with glassy carbon crucible

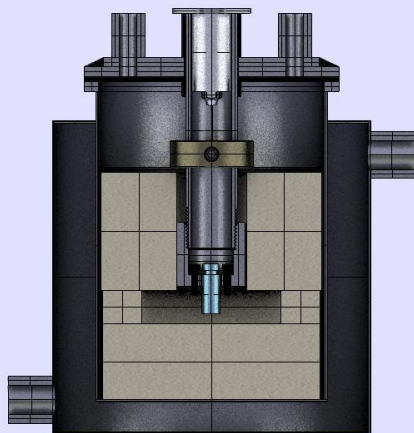


Containment vessel (6" I.D. x 8" Tall)
with cooling Jacket





Containment System for Heating and Off-gassing



Heating Element:

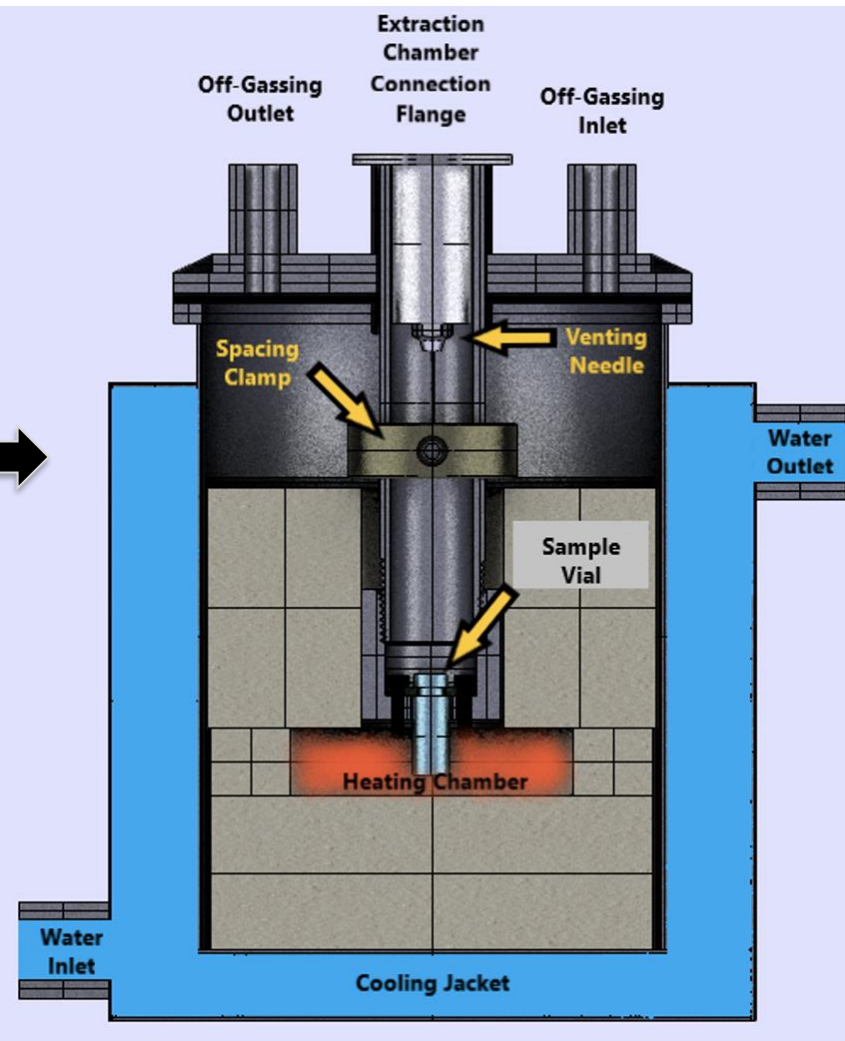
Ultralite kiln capable of reaching 850°C

Sample Vial:

Fused Silica glass 2 mL / 20 mL capacity

Mass of Sample:

MgCl₂, KCl, and UCl₃ = 3-20 g total





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Thanks to Dr.
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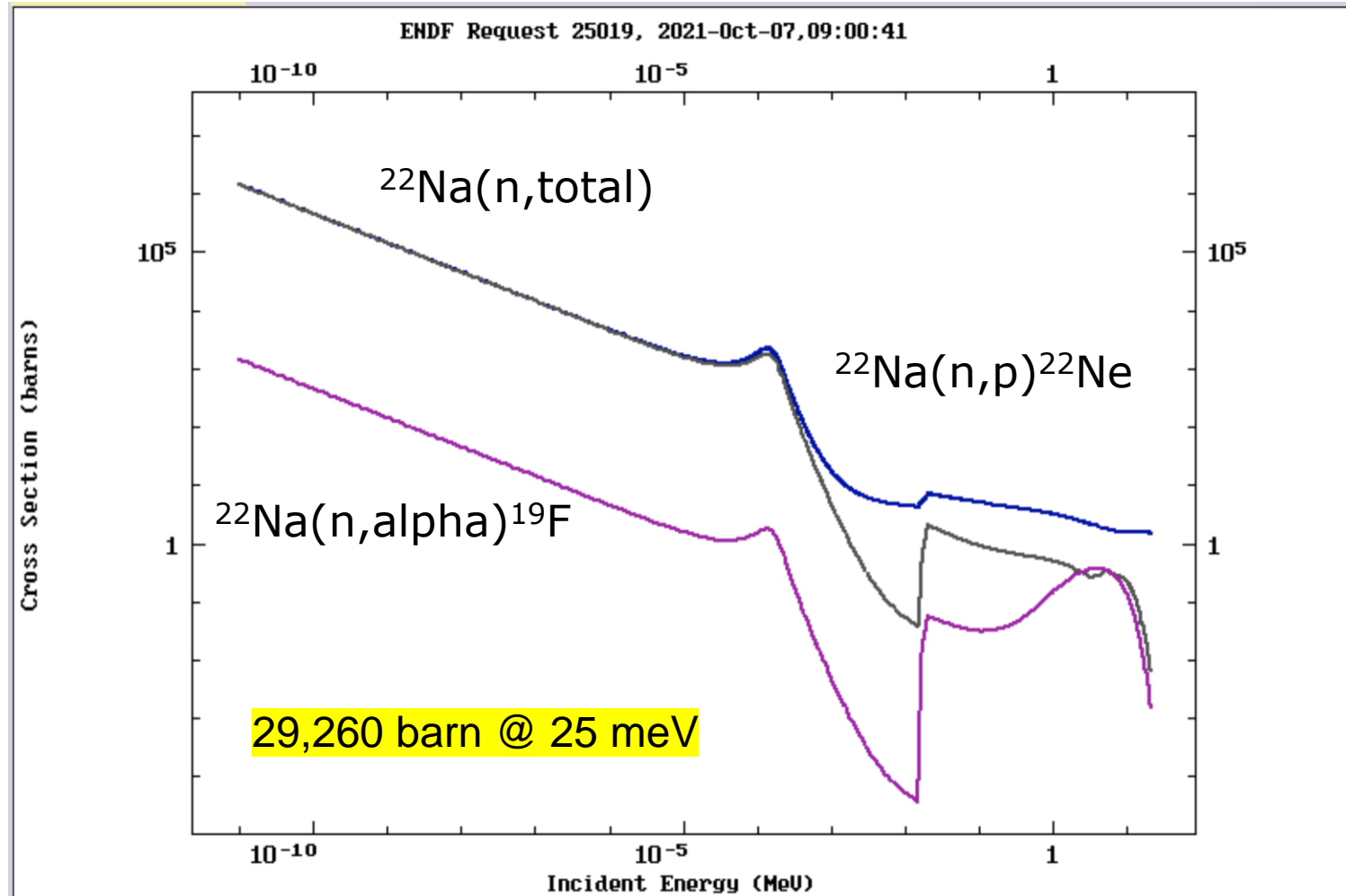


Thank you for your attention!



Challenges with tracer burn-up

- Na-22 has a significant neutron absorption cross-section





Radioisotope Selection

	Na-22	Co-60	Na-24	Br-82
Decay mode	β^+	β^-	β^-	β^-
Main gamma energy/keV	1274.54	1173.23; 1332.50	1368.63; 2754.01	554.35; 619; 776.52; 1044
Half life	2.6018 y	1925 d	14.977 h	35.28 h
Comments	Not a fission product	Selective bounding?	$T_{1/2}$ too short	$T_{1/2}$ too short



Collaboration with Dr. Shayan Shahbazi, Dr. Tingzhou Fei at ANL for burn-up simulation

20 grams of $\text{UCl}_3 + \text{KCl} + \text{MgCl}_2$

10% - 20% of U

2-4 grams of U

0.3% U-235 in DU

6 mg - 12 mg of U-235

Total volume of 2.0 cm^3

$\phi_{\text{th}} (\text{cm}^{-2} \cdot \text{s}^{-1})$	1E+13	Thermal Neutron Flux in CIF
$m_{\text{U235}} (\text{g})$	0.1	mass of U-235 in grams
$t_{\text{irr}} (\text{h})$	1	irradiation time in hours

^{133}Xe ; 5.25 d; 81 keV

Yield of ^{133}Xe at $t=0$, 14,700 μCi . $3.5\text{E}20$ atoms